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(Article begins on next page)

Cold brewing of rooibos tea affects its sensory profile and physicochemical properties compared to regular hot-, and boiled-brewing

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1 **ABSTRACT**

2 Cold-brewing of rooibos tea has gained popularity, in particular in Japan, one of its
3 major markets, due to the convenience of preparation. The sensory profile of traditional
4 (‘fermented’) rooibos, prepared by regular brewing (5 min infused in freshly-boiled
5 water) and served hot, is well described. However, no sensory profiling of cold-brewed
6 rooibos tea, rooibos tea served at ambient temperature or green rooibos tea has been
7 performed to date. In this study, the sensory profile of both fermented and green rooibos
8 prepared by cold brewing (8 h at ≤ 5 °C), regular brewing and boiling (5 min), and
9 consumed at ambient temperature (21 °C), were investigated. ‘Rooibos woody’ and
10 ‘fynbos-floral’ aroma notes of fermented rooibos were unaffected by brewing
11 procedure, but its cold brew was sweeter and less astringent than the other brews. The
12 sensory profile of green rooibos, dominated by a ‘hay/dried grass’ aroma, bitter taste
13 and astringency, was affected to a greater extent by brewing procedure, but mostly, cold
14 and regular brews had similar aroma attribute intensities. Higher levels of flavonoids
15 may explain the trends for astringency. Fe and Al were present in the highest levels in
16 boiled brews, although at levels too low to impact on health.

17

18 **Keywords**

19 Rooibos; cold brew; sensory profiling; flavonoids; aspalathin; Fe and Al content

20

21 **1. Introduction**

22 Sugar-sweetened beverages, including ready-to-drink teas, come at a price, not
23 only in terms of ‘sugar-tax’, but also the socio-economic burden of non-communicable
24 diseases. These beverages make a substantial contribution to sugar intake and have
25 consequently been linked to the onset of obesity, diabetes, and heart diseases (Benade &
26 Essop, 2017). In response, the beverage industry has reformulated beverages with lower
27 sugar content and/or replaced sugar with non-nutritive sweeteners. A recent report
28 predicted huge growth in the global sugar-free tea market by 2025 with the increasing
29 consumer awareness towards healthy drinks and a growing acceptance of natural herbs
30 identified as some of the market drivers ([https://www.openpr.com/news/2015719/sugar-](https://www.openpr.com/news/2015719/sugar-free-tea-market-to-see-huge-growth-by-2025-unilever)
31 [free-tea-market-to-see-huge-growth-by-2025-unilever](https://www.openpr.com/news/2015719/sugar-free-tea-market-to-see-huge-growth-by-2025-unilever)).

32 Another market trend is the emergence of cold-brewed tea, which could boost
33 the acceptance of sugar-free tea. Cold brewing delivers a beverage with **less caffeine**
34 and subsequently a reduced bitter taste as demonstrated for green tea (Lin, Liu, & Mau,
35 2008; Lin et al., 2014). A caffeine-free herbal tea that has grown in popularity on the
36 global market is rooibos (*Aspalathus linearis*) (Joubert & De Beer, 2011). The main
37 product is ‘fermented’ (oxidized) rooibos with its natural, slightly sweet taste and mild
38 astringency. The characteristic sweet-associated aroma notes, ‘caramel’ and ‘honey’, of
39 hot-brewed fermented rooibos (Jolley, Van der Rijst, Joubert, & Muller, 2017) could
40 even enhance the perception of sweetness through cross-modal aroma-taste interactions
41 (Alcaire, Antúnez, Vidal, Giménez, & Ares, 2017). In contrast to fermented rooibos,
42 green (unoxidized) rooibos is a relatively new product, developed to capitalize on the
43 market opportunities for antioxidant-rich beverages. High levels of aspalathin, a potent

44 C-glucosyl dihydrochalcone antioxidant, can be found in the unoxidized plant material
45 (Joubert & De Beer, 2011). Mounting evidence of aspalathin and green rooibos extracts
46 in the prevention of the metabolic syndrome (Johnson et al., 2018; Muller et al., 2018)
47 merits the development of green rooibos as a functional beverage. However, a ready-to-
48 drink beverage containing aspalathin-enriched green rooibos extract as an ingredient
49 was perceived as having an overt ‘plant-like’ character, disliked by consumers (Viljoen,
50 Muller, De Beer, & Joubert, 2017). Studies on functional foods have demonstrated that
51 consumers are not inclined to compromise on flavor and taste for health (Verbeke,
52 2006). Changing the brewing procedure of green rooibos may, therefore, be merited to
53 improve flavor and taste. Brewing temperature and duration of green tea were found to
54 affect flavor (Lee & Chambers, 2009) and the extraction of health-promoting substances
55 (Pastoriza, Pérez-Burillo, & Rufián-Henares, 2017). Cold-brewing of fermented rooibos
56 during summer has already gained acceptance in Japan, one of its main markets, due to
57 convenience. Adding a rooibos tea bag to water and infusing it for a prolonged period at
58 ambient temperature or in a refrigerator is a convenient alternative to steeping in hot
59 water or boiling on a stove. The latter practice was traditionally used to prepare rooibos
60 tea (Joubert, Gelderblom, Louw, & De Beer, 2008).

61 Against this background, the aim of this study was to determine the effect of
62 brewing procedure on the sensory profile and composition of both fermented rooibos
63 and green rooibos. The sensory profile, color, and turbidity of cold-brewed rooibos were
64 therefore compared with that of regular- and boiled-brewed rooibos. Color and turbidity
65 are important parameters for visual appeal. The chemical composition of the beverages
66 was quantified in terms of their Z-2-(β-D-glucopyranosyloxy)-3-phenylpropenoic acid

67 (PPAG), flavonoid, Al, and Fe content. PPAG was previously demonstrated to elicit a
68 bitter taste (Joubert et al., 2013).

69

70 **2. Experimental**

71 *2.1. Chemicals*

72 Reagent-grade chemicals were obtained from Sigma-Aldrich (St. Louis, MO,
73 USA) or Merck (Darmstadt, Germany). Ultrapure water for HPLC and metal analysis
74 was prepared by a two-stage water purification system (Elix and Milli-Q) (Merck).
75 Authentic standards (purity $\geq 95\%$) were purchased from Extrasynthese (Genay, France;
76 isoorientin, orientin, 7- β -D-glucopyranosyleryiodictyol (miscanthoside), isovitexin and
77 hyperoside), Carl Roth (Karlsruhe, Germany; vitexin), Sigma-Aldrich (isoquercitrin,
78 ferulic acid), Transmit (Gießen, Germany; rutin), Phytolab (Vestenbergsgreuth,
79 Germany; aspalathin (91% purity)) and SAMRC (Cape Town, South Africa;
80 nothofagin). PPAG was from our compound library (Plant Bioactives Group, ARC
81 Infruitec-Nietvoorbij). Stock solutions of the standards (ca. 1 g/L) were prepared in
82 DMSO and diluted with water as required. The diluted standard mixtures contained ca.
83 10 g/L ascorbic acid (Sigma-Aldrich) and were filtered through 0.45 μ m hydrophilic
84 PVDF filters (Merck).

85 Working standard solutions for metal analysis were prepared by appropriate
86 dilution of inorganic atomic absorption standard solutions (1000 mg/L) of Fe (Carlo
87 Erba, Milan, Italy) and Al (Sigma-Aldrich) and stored in a refrigerator at ca. 4 °C
88 protected from light. Ultrapure nitric acid (670-690 g/kg) was purchased from Carlo
89 Erba.

90 *2.2. Rooibos samples*

91 Samples from production batches (n=12) of both fermented rooibos and green
92 rooibos were sourced from Bokkeveld Rooibos (Nieuwoudtville, Northern Cape, South
93 Africa) and Rooibos Ltd. (Clanwilliam, Western Cape, South Africa), respectively.
94 Samples were mechanically sieved to remove dust (<0.42 mm) and coarse material
95 (>1.68 mm) (Jolley et al., 2017) and then stored at 21 °C in sealed glass jars.

96 *2.3. Preparation of infusions*

97 A cold brew (C), regular brew (R), and boiled brew (B) were prepared from each
98 batch of fermented and green rooibos by adding 12.5 g plant material to 1 L of distilled
99 water (equivalent to ‘cup-of-tea’ strength) as outlined in Fig. 1. Cold brewing entailed
100 adding water (21 °C) to the plant material in a wide-necked Schott bottle and steeping
101 for 8 h in a refrigerator (0-5 °C). For regular brew, freshly-boiled water was added to
102 the plant material in a glass jug followed by stirring for 5 s and steeping for 5 min
103 (Koch, Muller, Joubert, Van der Rijst, & Næs, 2012). The boiled brew was prepared by
104 adding boiling water to the plant material in a glass jug followed by microwave heating
105 (1000 W) for 5 min with stirring every minute to prevent boiling over. All brews were
106 strained through a fine-meshed strainer into 1 L Schott bottles directly after brewing and
107 stored overnight in a refrigerator (12 h). The brews were equilibrated to 21 °C prior to
108 sensory analysis.

109 *2.4. Descriptive sensory analysis*

110 Panel training and testing were performed separately for fermented and green
111 rooibos. Eleven assessors, with extensive experience in DSA of fermented rooibos,
112 were trained according to a basic protocol (Koch et al., 2012) to generate appropriate
113 aroma, flavor, taste, and mouthfeel attributes for green and fermented rooibos
114 (supplementary material, Table A.1). Attribute intensities were scored, using an

115 unstructured line scale (0 = not detected; 100 = high intensity), and the scores were
116 captured electronically (Compusense®, Guelph, Canada). Three batches of plant
117 material of each of the brew variants were analyzed per day in three consecutive
118 sessions. The sample order was randomized per assessor per session. The blind-coded
119 samples were served in black wine tasting glasses (ca. 70 mL per glass) covered with
120 plastic lids. Analyses were conducted in individual tasting booths under standard
121 lighting and controlled temperature (21 °C).

122 *2.5. Physicochemical analyses*

123 Turbidity, objective color, and soluble solids (SS) content of the brews were
124 determined in triplicate as described by Sishi, Muller, De Beer, Van der Rijst, & Joubert
125 (2019).

126 The concentration of the major flavonoids and PPAG in the brews was
127 determined in duplicate by high performance liquid chromatography with diode array
128 detection (HPLC-DAD) using an Agilent 1200 HPLC system (Agilent Technologies,
129 Santa Clara, CA, USA). All compounds, except aspalathin and ferulic acid, were
130 quantified using the validated method described by Walters, De Villiers, Joubert, & De
131 Beer (2017). Aspalathin and ferulic acid were quantified after separation on a Poroshell
132 120 EC-C₁₈ column (50×3 mm, 2.7 µm particle size; Agilent Technologies) at 25 °C.
133 The mobile phases (0.1% aqueous formic acid (A) and acetonitrile (B)) were run at 0.43
134 mL/min using the following gradient: 0–5 min, 12.4–14.5% B; 5–5.5 min, 14.5–80% B;
135 5.5–6.5 min, 80% B; 6.5–7 min, 80–12.4% B; 7–11 min, 12.4% B. UV-Vis spectra were
136 recorded between 200 and 450 nm with selected wavelength monitoring at 288 nm for
137 aspalathin and 320 nm for ferulic acid and the compounds quantified, using eight-point
138 calibration curves. Peak identity was verified by performing the same separation on a

139 Waters Acquity UPLC coupled to a Synapt G2 quadrupole-time-of-flight (Q-TOF) MS
140 detector, equipped with an electrospray ionization (ESI) source (Waters, Milford, USA).
141 The mass spectrometer was operated in negative MS^E mode with a collision energy
142 ramp from 20–60 V. All other MS parameters were as described by Walters et al.
143 (2017).

144 2.6. Quantification of Fe and Al

145 Samples were prepared and analyzed in triplicate in a clean room laboratory ISO
146 14644–1 Class 6, with areas at ISO Class 5 under laminar flow. All the laboratory
147 materials were acid-cleaned as described by Illuminati, Annibaldi, Truzzi, & Scarponi
148 (2014). An aliquot of each sample (5 mL) was put into Teflon vessels of a Microwave
149 Accelerated Reaction System, MARS-5, 1500 W (CEM, Mathews, NC, USA) and
150 digested (program reported in supplementary material, Table A.2) without any
151 pretreatment with 3 mL of HNO₃ (Annibaldi et al., 2019; Łozak, Sołtyk, Ostapczuk, &
152 Fijałek, 2002). An HP-500 control vessel containing the same matrix of samples was
153 used to control the temperature and pressure during the process. Water (2 mL) was used
154 to clean up the vessels at the end of the mineralization process and added to the
155 samples, to reach a final volume of 10 mL. The quantitative determinations of Al and Fe
156 were carried out using a graphite furnace atomic absorption spectrophotometer (DUO
157 240FS AA-GTA120 Graphite Tube Atomizer, Agilent, Santa Clara, CA, USA),
158 equipped with Zeeman background correction with argon (purity = 99.998%) as carrier
159 gas and multi-element hollow cathode lamps as light source. The instrumental
160 parameters used for each element are reported in the supplementary material (Table
161 A.3). The quantitative values were calculated from calibration curves generated using
162 diluted standard solutions (20–80 µg/L for Al and 10–50 µg/L for Fe).

163 For assessment of the accuracy of the data, Al and Fe were determined using a
164 certified reference material (dogfish muscle DORM-2, NRCC; Ottawa, ON, Canada).
165 Certified mean values and experimental mean values obtained from the analysis of
166 DORM-2, expressed in mg/kg dry weight (d.w.), are as follows: Fe, Certified value =
167 142 ± 10 , Analytical result found = 132 ± 1 ; Al, Certified value = 10.9 ± 1.7 , Analytical
168 result found = 10.2 ± 0.8 . No statistically significant differences ($P \geq 0.05$, Student's T-
169 test) between certified and measured values were detected, and results were in good
170 agreement with the certified values, proving good accuracy of the analytical
171 methodology.

172 2.7. *Statistical procedures*

173 Data obtained for fermented and green rooibos were analyzed separately. DSA
174 data were subjected to various statistical techniques to confirm panel reliability (Næs,
175 Brockhoff, & Tomic, 2010) and normality. Subsequent statistical analyses were
176 conducted on means over assessors. DSA and composition data were subjected to
177 analysis of variance (ANOVA) to test for treatment differences using SAS software
178 (Version 9.2, SAS Institute Inc., Cary, USA). Fisher's least significant difference (LSD)
179 was calculated (5% level) to compare treatment means. P values < 0.05 were considered
180 significant.

181 Principal component analysis (PCA), using the correlation matrix, was
182 performed, using XLStat (Version 2018, Addinsoft, Paris, France). Using the combined
183 fermented and green rooibos data sets, the Pearson correlation coefficient (r) was
184 calculated to determine the relationship between individual compound content and
185 bitterness and astringency, respectively.

186 3. Results and discussion

187 Twelve batches of both fermented and green rooibos were subjected to three
188 different brewing procedures, i.e. cold (C), regular (R), and boiled (B), to determine the
189 impact on their sensory profiles. As the mean intensity scores for the aroma and flavor
190 attributes followed similar trends with flavor attributes mostly perceived at lower
191 intensities, the focus will fall on the aroma.

192 3.1 Sensory profile

193 The PCA bi-plot for fermented rooibos (Fig. 2a) shows that only 52.7% of the
194 variance is explained. The regular- and boiled-brewed samples tend to associate on the
195 left of PC1, while the cold-brewed samples were positioned on the right, indicating that
196 the application of heat played a role in discriminating between treatments. On PC2,
197 samples largely separated between the boiled-brewed samples at the top and the cold-
198 and regular-brewed samples at the bottom. The intensities of the primary aroma
199 attributes ('rooibos-woody', 'fynbos-floral' and 'honey') of regular-brewed fermented
200 rooibos consumed hot (Jolley et al., 2017), as well as 'cooked apple', a secondary
201 aroma attribute of fermented rooibos, were not significantly ($P \geq 0.05$) affected by
202 treatment (Fig. 3a). The boiled-brewed samples associated with 'raisin' and 'caramel'
203 (Fig. 2a), which were perceived at significantly higher ($P < 0.05$) intensities in these
204 brews (Fig. 3a). The cold-brewed samples associated with 'tobacco' and sweet taste, but
205 not with astringency (Fig. 2a). The cold-brewed samples were significantly sweeter and
206 less astringent ($P < 0.05$) than the regular- and boiled-brewed samples (Fig. 3a), which
207 could contribute to consumer acceptance of cold-brewed rooibos. In view of the need to
208 reduce the sugar content of beverages to limit sugar intake, the inherent slightly sweeter
209 taste of cold-brewed rooibos is beneficial. The 'tobacco' aroma note is generally

210 associated with the dry ‘fermented’ leaf product. Furthermore, ‘seaweed’, a negative
211 aroma note, was not perceptible in the cold- and boiled-brewed samples, and just
212 perceptible in the regular brews ($P < 0.05$).

213 For green rooibos, PC1 and PC2 explained 77.6% of the variance and a clear
214 distinction between the samples of the three brewing procedures is evident (Fig. 2b). On
215 PC1, the boiled-brewed samples clustered on the right, whereas the cold- and regular-
216 brewed samples formed separate clusters on the left. On PC2, the separation between
217 samples was driven by brewing temperature with cold-brewed samples positioned at the
218 top and both regular- and boiled-brewed samples at the bottom. The cold brews had
219 significantly higher intensities ($P < 0.05$) of the aroma notes ‘fresh apple’ and ‘tobacco’,
220 as well as sweet taste, compared to those of the other brew types (Fig. 3b). In contrast,
221 the boiled brews had significantly higher intensities of the attributes ‘oats
222 porridge/cooked grains’, ‘caramel’, ‘stewed fruit’, bitter taste and astringency ($P < 0.05$)
223 than the other two treatments, whereas the negative attributes, ‘green grass’ and ‘putty’,
224 were barely perceptible in the boiled brews ($P < 0.05$) (Fig. 3b). The ‘seaweed’ aroma
225 note was prominent in the regular brews (intensity > 15), barely perceptible in the cold
226 brews (< 5), and practically not perceptible in the boiled brews. The trend is the same as
227 for fermented rooibos, suggesting that the short heat exposure is required to form
228 volatile compound(s) responsible for this negative aroma note, while a more severe
229 application of heat resulted in sufficient loss of the formed volatile compound(s).

230 The fermented and green rooibos samples did not originate from the same batches of
231 fresh plant material. However, the sensory profile of green rooibos was consistently and
232 vastly different to that of fermented rooibos, especially in that the vegetal (‘green grass’,
233 ‘cooked vegetables’) and cereal-like (‘oats porridge/cooked grains’) aroma attributes

234 were present in moderate to high intensities in green rooibos brews, depending on the
235 brewing procedure. While these aroma notes seem integral to the sensory profile of
236 green rooibos, they are considered to be negative for fermented rooibos (Jolley et al.,
237 2017). The ‘hay/dried grass’ intensity was not affected by the brewing procedure. A
238 short steam treatment of the dry fermented rooibos leaves also did not affect the ‘hay-
239 like’ intensity of rooibos infusions (regular-brewed; consumed hot) (Koch, Muller, De
240 Beer, Næs, & Joubert, 2013). Fermented rooibos brews had a sweeter taste, while the
241 green rooibos brews were more astringent with a prominent bitter taste.

242 3.2 SS content, color and turbidity

243 The SS content and turbidity of the fermented rooibos brews were significantly
244 ($P<0.05$) affected by brewing procedure, increasing progressively from cold to regular
245 to boiled brew (Table 1). The CIELab color parameters, lightness (L^*), and hue (h)
246 decreased and redness (a^*) increased significantly ($P<0.05$) (Table 1) with the increase
247 in the soluble solids content of the brew. The yellowness parameter, b^* , and
248 subsequently also chroma of the cold brews were significantly ($P<0.05$) lower than
249 those for the regular and boiled brews. Whilst the color parameters indicate differences
250 between the brews, the values are difficult to relate to the actual visual color in a cup or
251 glass due to dichroism, i.e. the hue or color depends on the degree of dilution or
252 container size (Francis & Clydesdale, 1975).

253 The different procedures used for the brewing of green rooibos had similar
254 effects to fermented rooibos on the soluble solids content, turbidity, L^* , a^* and hue of
255 the brews ($P<0.05$) (Table 2). For all these parameters, except a^* , higher values were
256 obtained for the green rooibos brews than for the fermented rooibos brews (Tables 1 and

257 2). The lower a^* values of green rooibos brews (1.0-14.3 vs 16.01-30.9) were to be
258 expected as the plant material is not oxidized.

259 3.3 PPAG and flavonoid content

260 The PPAG concentration in the fermented rooibos brews was slightly increased
261 ($P < 0.05$) by boiling compared to cold and regular brewing (Table 1; **representative**
262 **chromatograms shown in Fig. A.1**). The concentration of most of the flavonoids
263 increased significantly ($P < 0.05$) when heat was applied during **the** brewing of fermented
264 rooibos, with the boiling procedure being most effective in extracting the compounds
265 from the plant material (Table 1). The aspalathin, bioquercetin, hyperoside, and
266 isoquercitrin concentration of the regular and boiled brews did not differ significantly
267 ($P \geq 0.05$).

268 For the green rooibos brews, most phenolic compounds showed a significantly
269 ($P < 0.05$) higher concentration in the boiled brew compared to the other brews (Table 2;
270 **representative chromatograms shown in Fig. A.2**). Some exceptions were PPAG,
271 bioquercetin, and rutin with the lowest ($P < 0.05$) concentration in the regular brew. The
272 PPAG concentration of the cold and boiled brews was not significantly ($P \geq 0.05$)
273 different, **but the bioquercetin and rutin concentrations were significantly ($P < 0.05$)**
274 **higher in the boiled brew than in the cold brew**. Our previous study on the same sample
275 set provided more insight into the phenolic profile of the brews. A large number of
276 compounds were putatively annotated (Damiani et al., 2019). Of 187 putatively
277 annotated compounds, 32 and 35 polyphenols were considered as markers that
278 discriminate between cold and boiled brews, and cold and regular brews, respectively.

279 The investigation of the association between the flavonoid content of rooibos
280 infusions and the intensities of the taste modalities and astringency (when served hot) is

281 limited (Koch et al., 2013). The association between PPAG and sweet taste was notable,
282 however, when tested at 21 °C as a single compound, dissolved in water, it was bitter at
283 a concentration similar to that of regular-brewed rooibos. The PPAG concentration of
284 the brews analyzed for the present study (Tables 1 and 2) was much higher than its
285 detection threshold in water (0.4 mg/L) (Joubert et al., 2013). PPAG could, therefore, be
286 expected to contribute to the bitter taste. Considering that the green rooibos brews were
287 substantially more bitter than the fermented rooibos brews, while their respective PPAG
288 content did not differ to the same extent, other compounds or interaction with other
289 compounds (i.e. masking and/or enhancing effects) contributed to bitterness. A
290 moderate correlation was found between **the content of the individual flavonoids and the**
291 **bitter intensity of the brews** ($r \geq 0.5$; $P \leq 0.002$). The eriodictyol glucosides had the highest
292 correlation with bitter intensity ($r > 0.65$; $P < 0.0001$).

293 It is well-known that phenolic compounds contribute to **the** astringency of food,
294 wine, and tea (**Bordenave, Hamaker, & Ferruzzi, 2014**). Despite the green rooibos
295 brews having a **substantially** higher flavonoid content than the fermented rooibos brews,
296 the green rooibos brews were not substantially more astringent (Tables 1 and 2; Fig. 3).
297 A moderate, but significant correlation was found between **the** individual flavonoid
298 content of the brews and astringency ($r \geq 0.41$; $P < 0.01$). PPAG did not correlate
299 significantly with astringency ($r = 0.325$; $P = 0.053$).

300 3.4 Al and Fe content

301 Al and Fe were present in significantly higher ($P < 0.05$) levels in the boiled
302 brews of both fermented and green rooibos (Tables 1 and 2), while no significant
303 differences ($P \geq 0.05$) between the cold and regular brews were found, except in the case

304 of the Fe content of fermented rooibos brews. In this case, the Fe content increased
305 significantly ($P < 0.05$) from cold to regular to boiled brew.

306 The Fe and Al content values were similar to those previously reported for hot
307 water infusions of fermented rooibos (Joubert et al., 2008; Malik, Szakova, Drabek,
308 Balik, & Kokoska, 2008). Consumption of one cup of the brews would contribute
309 $< 0.2\%$ of the nutrient reference value (NRV) for adults, i.e. 18 mg/day, according to
310 South African food labeling regulations. Chronic Al intake, on the other hand, has
311 negative health implications (Pérez-Granados & Vaquero, 2002). A tolerable weekly
312 intake (TWI) of Al is 1 mg/kg body weight/week (Aguilar et al., 2008), i.e. 10 mg/day
313 by a 70 kg adult. One cup of rooibos per day would contribute $< 0.3\%$ of this intake.

314 3.5 “To cold-brew or not to cold-brew”

315 Boiling increased the flavonoid content of rooibos tea, irrespective of the type.
316 Similarly, in our previous study boiling also delivered brews with the highest
317 antioxidant capacities (ORAC and DPPH) (Damiani et al., 2019). However, it is not
318 clear whether these increases would translate into a specific health benefit. No
319 recommended daily intake values have been established for flavonoids, as this group of
320 phytochemicals differs vastly in terms of bioavailability and bioactivity, while the value
321 of antioxidant capacity in the assessment of health promotion is the subject of debate
322 (De Camargo et al., 2019; Williamson, Kay, & Crozier, 2018).

323 The flavor of beverage products is regarded as one of the main drivers of
324 consumer preference. The sensory profile associated with the respective rooibos brews
325 would be the deciding factor if convenience is not considered important. Cold brewing
326 of fermented rooibos delivered a similar primary and secondary sensory profile as
327 regular brewing, thus cold brewing offers no advantage or disadvantage in terms of its

328 sensory profile. The slightly higher intensities of ‘raisin’ and ‘caramel’ aroma notes of
329 boiled-brewed rooibos may be one of the reasons why rooibos was traditionally brewed
330 on a hot stove for extended periods. The ‘caramel’ note was also more prominent in
331 boiled-brewed green rooibos. The intensity of the ‘oats porridge/cooked grains’ note,
332 not present in fermented rooibos, was substantially increased with **the** boiling of green
333 rooibos. While the brewing procedure did not impact on astringency, boiling of green
334 rooibos increased the bitterness of the tea, linked to increased extraction of flavonoids.
335 To date, no study has been conducted to determine whether rooibos consumers are
336 willing to forfeit taste for health. Even the Fe and Al content of a cup of rooibos tea,
337 irrespective of type or brewing procedure, was found to be too low to impact,
338 respectively, positively and negatively on health and thus would not drive choice.

339 **4. Conclusions**

340 Cold brewing of both fermented rooibos and green rooibos did not offer **many**
341 **advantages** in terms of flavonoid content and sensory profile of the brew compared to
342 regular brewing. Notable was a less bitter and/or sweeter taste of the cold brew
343 compared to the boiled brew.

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354 **Conflict of interest**

355 The authors declare no conflict of interest.

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456

457

458

459 **Figure captions**

460 **Fig. 1.** Basic procedure used for the preparation of cold-, regular-, and boiled-brewed
461 fermented and green rooibos herbal tea.

462 **Fig. 2.** PCA bi-plot of the sensory profile of cold-, regular-, and boiled-brewed
463 fermented (a) and green (b) rooibos herbal tea with brewing procedure indicated by C,
464 R, and B, respectively. The numbers (1 to 12) refer to the individual samples,
465 representing independent replicates.

466 **Fig. 3.** ANOVA results of selected sensory attributes of cold-, regular-, and boiled-
467 brewed fermented (a) and green (b) rooibos herbal tea. Bars colored white, grey and
468 black represent cold-, regular-, and boiled-brewed rooibos, indicating mean values and
469 error bars (standard deviation).

470

471

COLD (C) BREW

1000 mL
ca 21°C water



8 h
@ 0-5 °C

12.5 g Rooibos

REGULAR (R) BREW

1000 mL
100 °C water

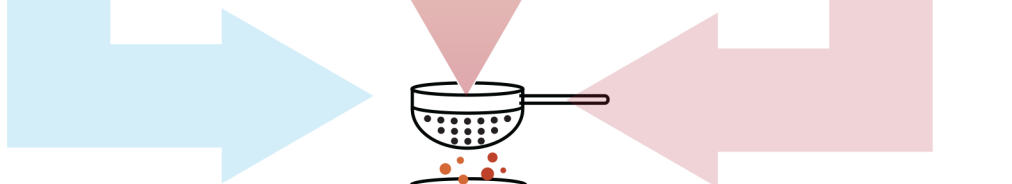


5 min
@ ca 21°C

12.5 g Rooibos

BOILED (B) BREW

1000 mL
100 °C water



Cooling to ca 21°C
Sensory and instrumental analyses

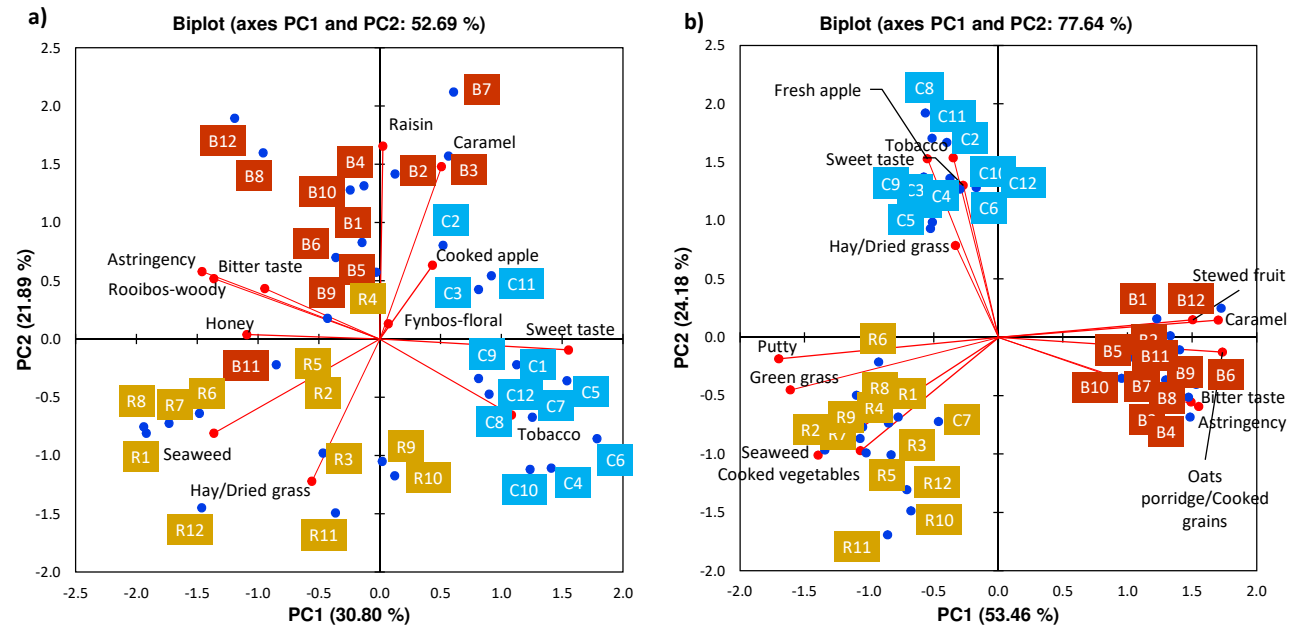


Figure 2

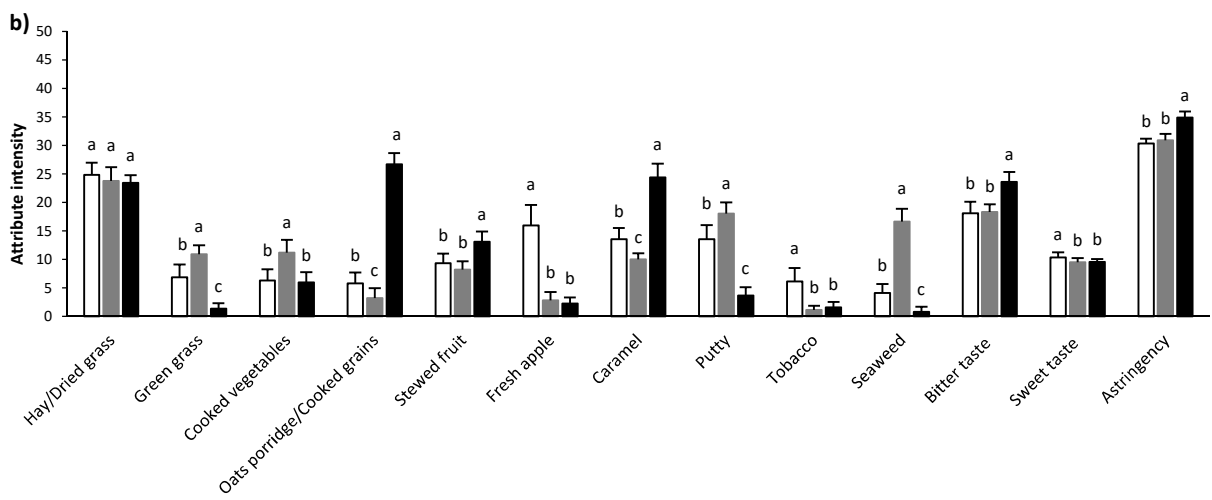
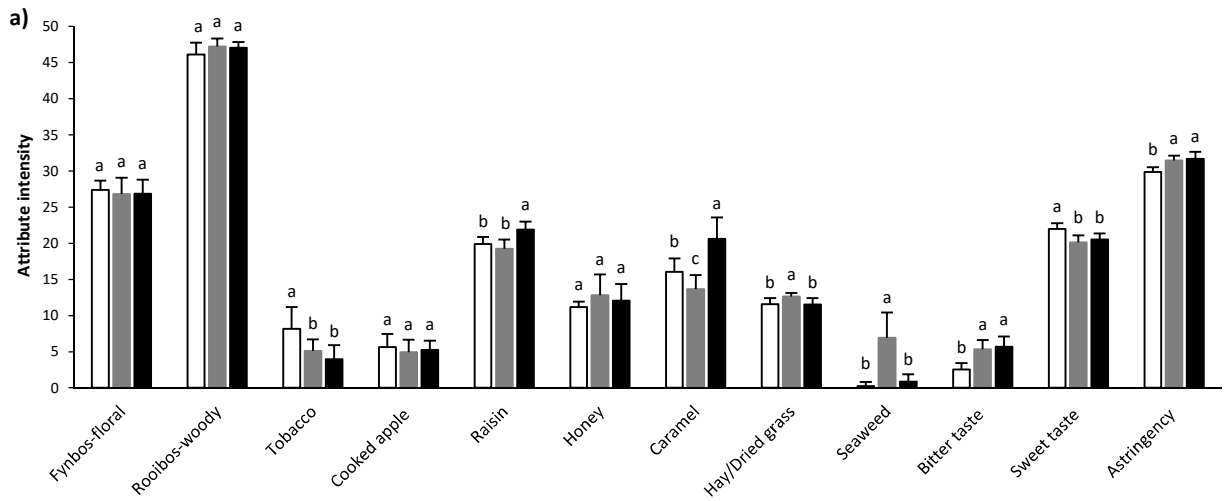


Figure 3

Table 1

Colour, turbidity (NTU), soluble solids content (SS, g/L) and chemical composition (mg/L, except Fe and Al, $\mu\text{g/L}$) of cold brew, regular brew and boiled brew fermented rooibos.

Parameter	Cold brew	Regular brew	Boiled brew	LSD
L*	81 \pm 2 a	72 \pm 2 b	64 \pm 3 c	1
a*	16 \pm 3 c	26 \pm 2 b	31 \pm 2 a	1
b*	93 \pm 5 b	97 \pm 3 a	95 \pm 3 a	2
C	94 \pm 5 b	101 \pm 3 a	100 \pm 3 a	2
h	80 \pm 1 a	75 \pm 1 b	72 \pm 1 c	1
Turbidity	5 \pm 1 c	22 \pm 7 b	42 \pm 14 a	6
Soluble solids	1.0 \pm 0.1 c	1.5 \pm 0.2 b	1.9 \pm 0.2 a	0.1
SE6G ^a	7.3 \pm 0.7 c	10.8 \pm 1.0 b	14.0 \pm 1.4 a	0.5
RE6G ^b	6.9 \pm 0.8 c	10.5 \pm 1.0 b	13.8 \pm 1.5 a	0.5
SE8G ^c	2.0 \pm 0.3 c	3.0 \pm 0.4 b	4.3 \pm 0.7 a	0.2
RE8G ^d	2.5 \pm 0.2 c	3.4 \pm 0.3 b	4.7 \pm 0.5 a	0.2
Aspalathin	3 \pm 2 b	5 \pm 4 a	5 \pm 3 a	1
PPAG ^e	9 \pm 1 b	8 \pm 1 b	10 \pm 2 a	0.4
Nothofagin	0.3 \pm 0.2 c	0.6 \pm 0.3 b	0.9 \pm 0.4 a	0.1
Isoorientin	4.4 \pm 0.6 c	11.3 \pm 1.4 b	15.9 \pm 1.3 a	0.6
Orientin	8 \pm 1 c	18 \pm 2 b	25 \pm 2 a	1
Ferulic acid	1.2 \pm 0.4 c	1.3 \pm 0.3 b	1.6 \pm 0.4 a	0.1
Bioquercetin	5 \pm 3 b	9 \pm 4 a	10 \pm 3 a	1
Vitexin	1.3 \pm 0.1 c	3.1 \pm 0.3 b	4.1 \pm 0.3 a	0.2
Hyperoside	0.9 \pm 0.9 b	2.4 \pm 1.8 a	2.4 \pm 0.9 a	0.5
Rutin	0.8 \pm 0.6 c	1.5 \pm 1.0 b	2.1 \pm 1.4 a	0.5
Isovitexin	1.3 \pm 0.1 c	3.2 \pm 0.3 b	4.4 \pm 0.3 a	0.2
Isoquercitrin	0.3 \pm 0.4 b	1.8 \pm 1.4 a	1.9 \pm 1.0 a	0.6
Luteoleoside	0.02 \pm 0.04 c	0.51 \pm 0.18 b	0.94 \pm 0.22 a	0.12
Sum of flavonoids	45	85	111	-
Fe	105 \pm 13 c	119 \pm 14 b	142 \pm 5 a	7
Al	42 \pm 22 b	57 \pm 21 b	108 \pm 35 a	20

^a (S)-6- β -D-glucopyranosyleriodictyol.

^b (R)-6- β -D-glucopyranosyleriodictyol.

^c (S)-8- β -D-glucopyranosyleriodictyol.

^d (R)-8- β -D-glucopyranosyleriodictyol.

^e Z-2-(β -D-glucopyranosyloxy)-3-phenylpropenoic acid.

Table 2

Colour, turbidity (NTU), soluble solids content (SS, g/L) and chemical composition (mg/L, except Fe and Al, $\mu\text{g/L}$) of cold brew, regular brew and boiled brew green rooibos.

Parameter	Cold brew	Regular brew	Boiled brew	LSD
L*	93 \pm 1 a	88 \pm 3 b	80 \pm 2 c	1
a*	1 \pm 1 c	6 \pm 2 b	14 \pm 2 a	1
b*	40 \pm 7 b	35 \pm 6 c	46 \pm 5 a	3
C	40 \pm 7 b	35 \pm 7 c	49 \pm 5 a	3
h	89 \pm 1 a	80 \pm 2 b	73 \pm 1 c	1
Turbidity	9 \pm 3 c	33 \pm 10 b	60 \pm 9 a	5
Soluble solids	1.9 \pm 0.2 b	1.7 \pm 0.4 b	2.5 \pm 0.4 a	0.2
Aspalathin	189 \pm 27 b	176 \pm 48 b	267 \pm 40 a	17
PPAG ^a	13 \pm 2 a	9 \pm 2 b	13 \pm 1 a	1
Nothofagin	19 \pm 4 b	18 \pm 6 b	28 \pm 6 a	2
Isoorientin	13 \pm 2 b	13 \pm 3 b	21 \pm 3 a	1
Orientin	19 \pm 2 b	19 \pm 4 b	30 \pm 4 a	2
Bioquercetin	18 \pm 5 b	14 \pm 5 c	21 \pm 6 a	1
Vitexin	3.0 \pm 0.4 b	2.8 \pm 0.7 b	4.5 \pm 0.6 a	0.3
Hyperoside	5 \pm 2 b	5 \pm 2 b	7 \pm 3 a	1
Rutin	8 \pm 2 b	6 \pm 2 c	10 \pm 2 a	1
Isovitexin	3.6 \pm 0.4 b	3.7 \pm 0.9 b	5.7 \pm 0.8 a	0.3
Isoquercitrin	8 \pm 3 b	8 \pm 4 b	13 \pm 5 a	1
Luteoleoside	0.7 \pm 0.3 c	1.1 \pm 0.4 b	1.9 \pm 0.6 a	0.2
Sum of flavonoids	286	267	409	-
Fe	21 \pm 8 b	21 \pm 7 b	53 \pm 14 a	6
Al	82 \pm 42 b	81 \pm 38 b	149 \pm 45 a	32

^a Z-2-(β -D-glucopyranosyloxy)-3-phenylpropenoic acid.



FERMENTED ROOIBOS



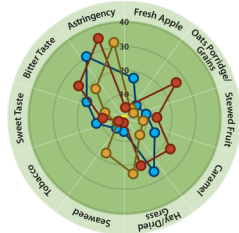
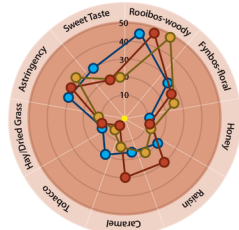
GREEN ROOIBOS



PREPARATION PROCEDURE



AROMA PROFILE



- C
- R
- B